

Chapter 20 Acids Bases Guided Reading Answers

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Figure 20.1 All of these items contain acids or bases, or produce acids or bases upon dissolving in water. Tomatoes contain ascorbic acid; tea contains tannic acid. The base calcium hydroxide is a component of mortar, which was used to build the Great Wall of China. Antacids use a variety of bases to neutralize excess stomach acid.

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Chapter 20 acids and bases. STUDY. PLAY. Hydrogen-ion donor. an acid, according to the Bronsted-Lowry theory. Hydrogen-ion acceptor. A base, according to the Bronsted-Lowry theory, ammonia acts as a base when it accepts hydrogen ions from water. Conjugate acid.

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So, whether acid or base are added they are removed to keep pH constant. The common ion effect is the lowering of the solubility of a substance by the addition of a common ion. This can be easily seen in + Acids and Bases Properties of Acids and Bases A. Properties of Acids 2) 3

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Brønsted-Lowry Acids & Bases Identify each species in the following equation as either the Brønsted-Lowry acid, the Brønsted-Lowry base, the conjugate acid, or the conjugate base. Identify the conjugate acid-base pairs in the reaction. $\text{H}_2\text{SO}_4(\text{aq}) + \text{HPO}_4^{2-}(\text{aq}) \rightarrow \text{HSO}_4^-(\text{aq}) + \text{H}_2\text{PO}_4^-(\text{aq})$ Strong vs. Weak Acids and Bases Strong ...

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