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### CHEMISTRY NOTES FOR FORM FOUR - MSOMI BORA

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$\text{NH}_4\text{OH}(\text{aq}) \rightarrow \text{NH}_4^+(\text{aq}) + \text{OH}^-(\text{aq})$   $\text{Ca}(\text{OH})_2(\text{aq}) \rightarrow \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^-(\text{aq})$   $\text{Mg}(\text{OH})_2(\text{aq}) \rightarrow \text{Mg}^{2+}(\text{aq}) + 2\text{OH}^-(\text{aq})$   
3. An acid is a proton donor. A base is a proton acceptor. this is called Bronsted-Lowry definition of acids and bases. From this definition, an acid donates  $\text{H}^+$  .

## Chemistry Notes Form 4 - Chemistry Form Four Pdf - Online ...

Short Notes: Form 4 Chemistry Chemical Formulae and Equation Calculation . Summary Summary ÷ molar volume × molar volume × molar mass . Avogadro Constant . Number of particles . ... The metal oxide of group 1 elements can dissolve in water to form alkali (hydroxide) solution .  $\text{Li}_2\text{O} + \text{H}_2\text{O} \rightarrow 2\text{LiOH}$  . Na. 2. O + H. 2. O  $\rightarrow 2\text{NaOH}$  . K. 2 ...

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notes form one.

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## **Chemistry Notes and Revision Questions and Answers, Form ...**

Chemistry Notes Form 1 . Chemistry. Form 1 Guide. 1.0.0 Introduction to Chemistry. Welcome to Chemistry, the Science concerned with the study of matter. In simplest terms, chemistry is the science of matter. Anything that can be touched, tasted, smelled, seen or felt is made of chemicals.

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## **SPM Form 4 Chemistry Chapter 4 - Periodic Table - SPM ...**

Short Notes: Form 4 Chemistry Chemical Formulae and Equation Calculation Summary Summary ÷ molar volume × molar volume × molar mass Avogadro Constant Number of particles The metal oxide of group 1 elements can dissolve in water to form ...

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[KBSM] Form 4 Chemistry Based on the SPM Form 4 Chemistry syllabus. In this subject, you will learn the structure of atom, chemical formulae and equations, periodic table of elements and chemical bonds. You will also learn about electrochemistry, acids and bases, salts and manufactured substances in industry.

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